



Standard Test Method for Determination of Asphaltenes (Heptane Insolubles) in Crude Petroleum and Petroleum Products¹²

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1. Scope

1.1 This test method covers a procedure for the determination of the heptane insoluble asphaltene content of gas oil, diesel fuel, residual fuel oils, lubricating oil, bitumen, and crude petroleum that has been topped to an oil temperature of 260°C (see A1.2.1.1).

1.2 The precision is applicable to values between 0.50 and 30.0 % m/m. Values outside this range may still be valid but may not give the same precision values.

1.3 Oils containing additives may give erroneous results.

1.4 The values stated in SI units are to be regarded as the standard.

1.5 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 ASTM Standards:

D 86 Test Method for Distillation of Petroleum Products³

D 1298 Test Method for Density, Relative Density (Specific Gravity), or API Gravity of Crude Petroleum and Liquid Petroleum Products by Hydrometer Method³

D 4052 Test Method for Density and Relative Density of Liquids by Digital Density Meter⁴

D 4057 Practice for Manual Sampling of Petroleum and Petroleum Products⁴

D 4177 Practice for Automatic Sampling of Petroleum and Petroleum Products⁴

¹ This test method is under the jurisdiction of ASTM Committee D02 on Petroleum Products and Lubricants and is the direct responsibility of Subcommittee D02.14 on Stability and Cleanliness of Liquid Fuels.

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² This standard is based on material published in the *IP Standard methods for Analysis and Testing of Petroleum and Related Products and British Standard 2000 Paris*, copyright The Institute of Petroleum, 61 New Cavendish Street, London, W1M 8AR. Adapted with permission of The Institute of Petroleum.

³ *Annual Book of ASTM Standards*, Vol 05.01.

⁴ *Annual Book of ASTM Standards*, Vol 05.02.

2.2 IP Standard:⁵

Specifications for IP Standard Thermometers

3. Terminology

3.1 Definitions:

3.1.1 *asphaltenes, n*—wax-free organic material insoluble in heptane, but soluble in hot toluene (benzene).

NOTE 1—Benzene is included in this definition solely on the basis of its classical references in the definition of asphaltenes. The precision of this test method when using toluene has been found to be the same as when using benzene.

4. Summary of Test Method

4.1 A test portion of the sample is mixed with heptane and the mixture heated under reflux, and the precipitated asphaltenes, waxy substances, and inorganic material are collected on a filter paper. The waxy substances are removed by washing with hot heptane in an extractor.

4.2 After removal of the waxy substances, the asphaltenes are separated from the inorganic material by dissolution in hot toluene, the extraction solvent is evaporated, and the asphaltenes weighed.

5. Significance and Use

5.1 Asphaltenes are the organic molecules of highest molecular mass and carbon-hydrogen ratio normally occurring in crude petroleum and petroleum products containing residual material. They may give problems during storage and handling if the suspension of asphaltene molecules is disturbed through excess stress or incompatibility. They are also the last molecules in a product to combust completely, and thus may be one indicator of black smoke propensity. Their composition normally includes a disproportionately high quantity of the sulfur, nitrogen, and metals present in the crude petroleum or petroleum product.

⁵ Available from Institute of Petroleum, 61 New Cavendish St., London, W1M 8AR, UK.

6. Apparatus

6.1 *General*—Ground-glass joints from different sources may have one of two diameter to length ratios. For the purposes of this test method, either is suitable, and for some applications, the diameter itself can be one of two. However, it is critical that the male and female parts of each joint are from the same series to avoid recession or protuberance.

6.2 *Condenser*, with a coil or double surface, fitted with a 34/45 or 34/35 ground-glass joint at the bottom to fit the top of the extractor. Minimum length is 300 mm.

6.3 *Reflux Extractor*, conforming to the dimensions given in Fig. 1. Tolerances are ± 1 mm on the height and outer diameter (OD) of the extractor body and ± 0.5 mm on all other dimensions. The female ground-glass joint at the top shall match the male at the bottom of the condenser, and the male ground-glass joint at the bottom shall match the female of the conical flask.

6.4 *Conical Flasks*, of borosilicate glass of appropriate capacity (see 11.2 and Table 1), with ground-glass joints to fit the bottom of the extractor.

NOTE 2—Sizes 24/39, 24/29, 29/43 or 29/32 are suitable.

6.5 *Stopper*, of borosilicate glass of a size to fit the conical flask.

6.6 *Evaporating Vessel*, of borosilicate glass. Either a hemispherical dish of approximately 90 mm diameter, or another suitable vessel used in conjunction with a rotovapor.

NOTE 3—A rotovapor in conjunction with a nitrogen atmosphere reduces the hazard of toluene evaporation (see 11.7).

6.7 *Filter Funnel*, of borosilicate glass, approximately 100 mm diameter.

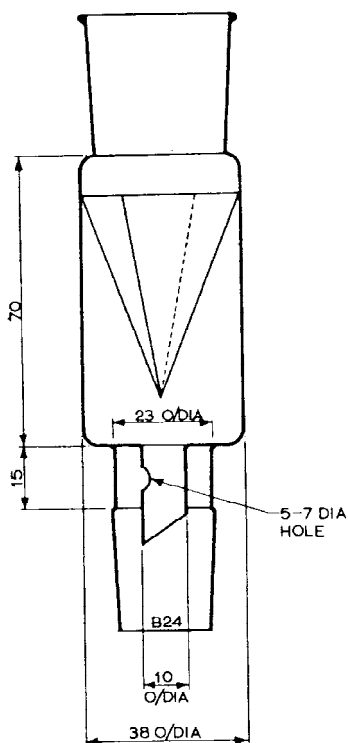


FIG. 1 Extractor

TABLE 1 Test Portion Size, Flask, and Heptane Volumes

Estimated Asphaltene Content % m/m	Test Portion Size g	Flask Volume mL	Heptane Volume mL
Less than 0.5	10 \pm 2	1000	300 \pm 60
0.5 to 2.0	8 \pm 2	500	240 \pm 60
Over 2.0 to 5.0	4 \pm 1	250	120 \pm 30
Over 5.0 to 10.0	2 \pm 1	150	60 \pm 15
Over 10.0 to 25.0	0.8 \pm 0.2	100	25 to 30
Over 25.0	0.5 \pm 0.2	100	25 \pm 1

6.8 *Filter Papers*, Whatman⁶ Grade 42, 110 or 125-mm diameter.

6.9 *Analytical Balance*, capable of weighing with an accuracy of 0.1 mg.

6.10 *Forceps*, of stainless steel, spade ended.

6.11 *Timing Device*, electronic or manual, accurate to 1.0 s.

6.12 *Oven*, capable of maintaining a temperature from 100 to 110°C.

6.13 *Graduated Cylinders*, of 50 and 100 mL capacity.

6.14 *Stirring Rods*, of glass or polytetrafluoroethylene (PTFE), 150 by 3-mm diameter.

6.15 *Cooling Vessel*, consisting of either a desiccator without desiccant, or another suitable tightly-stoppered vessel.

6.16 *Mixer*, high-speed, nonaerating.

7. Reagents

7.1 *Toluene (methylbenzene)* [C₆H₅CH₃], analytical reagent or nitration grade.

7.2 *Heptane* [C₇H₁₆], analytical reagent grade.

8. Sampling

8.1 Unless otherwise specified, take samples by the procedures described in Practice D 4057 or D 4177.

9. Test Portion Preparation

9.1 Test portions from the laboratory samples shall be drawn after thorough mixing and subdivision. Heat viscous samples of residual fuels to a temperature that renders the sample liquid, but not above 80°C, and homogenize, using the mixer (see 6.16) as necessary.

9.2 Heat samples of penetration grade bitumens to a temperature not exceeding 120°C, and stir well before taking an aliquot.

9.3 Samples of hard bitumens shall be ground to a powder before an aliquot is taken.

9.4 Samples of crude petroleum shall be prepared in accordance with the procedure described in the Annex A1, unless it is known that the crude petroleum contains negligible quantities of material boiling below 80°C.

10. Apparatus Preparation

10.1 Clean all glass flasks (see 6.4) and dishes (see 6.6) by a means that matches the cleanliness obtained by the use of a strongly oxidizing agent, such as chromosulfuric acid, ammonium peroxydisulfate in concentrated sulfuric acid at approximately 8 g/L, or sulfuric acid itself, soaking for at least 12 h,

⁶ Available from Whatman Int. Ltd., Maidstone, UK.

followed by rinses in tap water, distilled water, and then acetone, using forceps only for handling. (**Warning**—Chromosulfuric acid is a health hazard. It is toxic, a recognized carcinogen as it contains Cr (VI) compounds, which are highly corrosive and potentially hazardous in contact with organic materials. When using chromosulfuric acid cleaning solution, eye protection and protective clothing are essential. Never pipette the cleaning solution by mouth. After use, do not pour cleaning solution down the drain, but neutralize it with great care, owing to the concentrated sulfuric acid present, and dispose of it in accordance with standard procedures for toxic laboratory waste (chromium is highly dangerous to the environment). Nonchromium containing, strongly oxidizing acid cleaning solutions are also highly corrosive and potentially hazardous in contact with organic materials, but do not contain chromium, which has its own special disposal problems.

10.2 For routine analysis, use a proprietary laboratory detergent to clean the glassware, followed by the rinses described in 10.1. When the detergent cleaning no longer matches the cleanliness required based on visual appearance, use a strong oxidizing agent.

10.3 After rinsing, place the glassware in the oven (see 6.12) for 30 min, and cool in the cooling vessel (see 6.15) for 30 min before weighing.

11. Procedure

11.1 Estimate the asphaltene content of the sample, or residue obtained from the procedure in the Annex A1, and weigh the quantity, to the nearest 1 mg for masses above 1 g and to the nearest 0.1 mg for masses of 1 g and below (see Table 1), into a flask (see 6.4) of appropriate capacity (see Table 1).

11.2 Add heptane (see 7.2) to the test portion in the flask at a ratio of 30 mL to each 1 g of sample if the expected asphaltene content is below 25 % m/m. For samples with an expected asphaltene content of above 25 % m/m, a minimum heptane volume of 25 mL shall be used (see Table 1).

11.3 Boil the mixture under reflux for 60 ± 5 min. Remove the flask and contents at the end of this period, cool, close with a stopper (see 6.5), and store in a dark cupboard for 90 to 150 min, calculated from the time of removal from reflux.

11.4 Place the filter paper, folded as shown in Fig. 2 (so as to prevent loss of asphaltenes by creeping), in the filter funnel, using forceps. Thereafter, handle the filter paper only with forceps. Without agitation, decant the liquid into the filter paper, and then transfer the the residue in the flask as

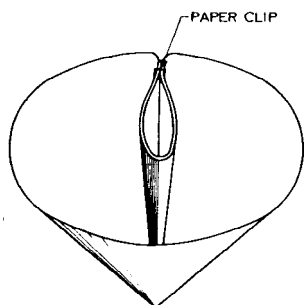


FIG. 2 Filter Paper

completely as possible with successive quantities of hot heptane, using the using the stirring rod (see 6.14) as necessary. Give the flask a final rinse with hot heptane, and pour the rinsings through the filter. Set the flask aside, without washing, for use as specified in 11.6.

11.5 Remove the filter paper and contents from the funnel, and place in the reflux extractor (see 6.3). Using a flask different from that used initially, reflux with heptane (see 7.2) at a rate of 2 drops/s to 4 drops/s from the end of the condenser for an extraction period of not less than 60 min, or until a few drops of heptane from the bottom of the extractor leave no residue on evaporation on a glass slide.

11.6 Replace the flask by the one used initially, and to which has been added 30 to 60 mL of toluene (see 7.1), and continue refluxing until all the asphaltenes have been dissolved from the paper.

11.7 Transfer the contents of the flask to a clean and dry (see Section 10) evaporating vessel (see 6.6), weighed to the nearest 0.2 mg by tare against a similar dish. Wash out the flask with successive small quantities of toluene to a total not exceeding 30 mL. Remove the toluene by evaporation on a boiling water bath (**Warning**—Perform the evaporation in a fume hood.), or by evaporation in a rotovapor under an atmosphere of nitrogen.

11.8 Dry the dish and contents in the oven (see 6.12) at 100 to 110°C for 30 min. Cool in the cooling vessel (see 6.15) for 30 to 60 min and reweigh by tare against the dish used previously for this purpose, and which has been subjected to the same heating and cooling procedure as was the dish containing asphaltenes.

NOTE 4—Asphaltenes are very susceptible to oxidation, and it is important that the procedure specified in the final drying stage is adhered to exactly as regards to temperature and time ± 1 min.

12. Calculation

12.1 Calculate the asphaltene content, A , in % m/m, of petroleum products as follows:

$$A = 100 (M/G) \quad (1)$$

where:

M = is the mass of asphaltenes, in grams, and

G = is the mass of test portion, in grams.

12.2 Calculate the asphaltene content, C , in % m/m, of crude petroleum prepared in the manner described in the Annex, as follows:

$$C = 100 (MR/GD) \quad (2)$$

where:

M = is the mass of asphaltenes, in grams,

R = is the mass of the residue from distillation, in grams,

G = is the mass of the residue aliquot, in grams, and

D = is the mass of crude petroleum sample distilled, in grams.

13. Report

13.1 Report the following information:

13.1.1 The heptane insoluble asphaltene content of values less than 1.00 % m/m, to the nearest 0.05 % m/m, by Test Method D 6560.

13.1.2 The heptane insoluble asphaltene content of values

of 1.0 % m/m and greater, to the nearest 0.1 % m/m, by Test Method D 6560.

14. Precision and Bias

14.1 *Precision*—The precision values were determined in an interlaboratory program, using benzene as solvent, in 1956. A second interlaboratory program, using toluene as solvent, was carried out in 1975 to confirm the precision. No data from the 1956 evaluation program can be found.

NOTE 5—A recent (1998) interlaboratory program in France on four samples with asphaltene contents in the range from 0.50 to 22.0 % m/m resulted in estimated precision values very similar to those given in 14.2 and 14.3, except for the reproducibility at very low levels (<2.0 % m/m), which was significantly worse.

14.2 *Repeatability*—The difference between two successive test results obtained by the same operator with the same apparatus under constant operating conditions on identical test material would, in the normal and correct operation of the test method, exceed the value below in only one case in twenty.

$$r = 0.1 A \quad (3)$$

where A is the average result, in % m/m.

14.3 *Reproducibility*—The difference between two single and independent results obtained by different operators working in different laboratories on nominally identical test material would, in the normal and correct operation of the test method, exceed the value below in only one case in twenty.

$$R = 0.2 A \quad (4)$$

where A is the average result, in % m/m.

14.4 *Bias*—Since heptane insoluble asphaltenes are defined by this test method, no bias can be assigned. The term asphaltenes may be associated with other similar procedures using alkanes as precipitating agents other than heptane. In general, lighter alkanes will give higher results and heavier alkanes will give lower results than heptane, but no consistent ratio can be assigned.

15. Keywords

15.1 asphaltenes; bitumen; crude petroleum; diesel fuel; gas oil; heptane insoluble residues; lubricating oil; residual fuel oils

ANNEX

(Mandatory Information)

A1. PREPARATION OF CRUDE PETROLEUM RESIDUE BY DISTILLATION

A1.1 Scope

A1.1.1 This annex covers a procedure for the preparation of a crude petroleum residue (topped to an oil temperature of 260°C) suitable for the determination of asphaltene content (see 9.4).

A1.2. Terminology

A1.2.1 Definitions:

A1.2.1.1 *crude petroleum residue, n*—residue from distillation to an oil temperature of 260°C carried out under conditions of this specific preparatory procedure.

A1.3 Summary of Test Method

A1.3.1 A weighed test portion of the crude petroleum sample is distilled in specified glass apparatus under specified conditions of heat input and rate of distillation. The distillation is stopped at an oil temperature of 260°C and the mass of the residue determined.

A1.4 Apparatus

A1.4.1 Distillation Apparatus:

A1.4.1.1 *Residue Distillation Apparatus*, conforming to the requirements of Test Method D 86, with the exception of the temperature sensor.

(a) The temperature sensor shall be a liquid-in-glass thermometer of total immersion type of total length 300 to 320 mm, with a temperature range from –4 to 360°C, and a maximum scale error of 2°C, or an alternative temperature measurement device or system of at least equivalent accuracy.

A transparent bath, maintained at $15 \pm 3^\circ\text{C}$, is mandatory for the receiver.

NOTE A1.1—A suitable thermometer is an IP 4C, as specified in Appendix A of the IP Test Methods Book.

A1.4.1.2 *Drying Distillation Apparatus*, conforming to A1.4.1.1, with the exception of a 500-mL distillation flask instead of the flask specified, and a 200-mL separating funnel immersed in broken ice instead of the receiver.

A1.4.2 *Drying Apparatus*, consisting of one of the following:

A1.4.2.1 *Centrifuge*.

A1.4.2.2 *Filter Apparatus*, operating at increased pressure.

A1.4.2.3 *Steel Container*, capable of withstanding 1000 kPa pressure and of being heated to 200°C.

A1.4.2.4 *Cottrell-type Electrostatic Separator*.

A1.5 Sample Preparation

A1.5.1 If the crude petroleum contains sufficient water to cause difficulties in distillation (excessive foaming), dry the sample by a method that avoids the loss of volatile components. The methods given in A1.5.1.1-A1.5.1.5 are suitable.

A1.5.1.1 Separate the water by gravity or by centrifuging in a closed container at the lowest practicable temperature.

A1.5.1.2 Filter the sample at the lowest practicable temperature and applied pressure, through anhydrous calcium chloride, sodium sulfate, or other suitable drying agent, in a closed vessel.

A1.5.1.3 Heat the sample in a closed steel container (see A1.4.2.3) fitted with a temperature sensor and pressure sensor.

Fill the container to approximately 70 % capacity, and heat it until the oil reaches a temperature of 200°C, or until the pressure reaches 700 kPa. Allow the container to cool to ambient temperature, and then decant the oil from the separated water.

A1.5.1.4 Construct a Cottrell-type separator from a tall glass beaker with a brass gauze cylinder, lined with flannel that has been saturated with water, and then squeezed out to leave it damp, fitted tightly inside. A brass gauze cylinder mounted on the glass spindle of a laboratory stirrer, rotated at approximately 30 rev/min, forms the central electrode. Pour the sample into the beaker, and apply a voltage to the electrodes. Interrupt the water precipitation at intervals to prevent the loss of volatile components, and allow the oil to cool. Separate the oil from the water that has coalesced and run down the flannel lining.

A1.5.1.5 Distil off the water with the more volatile hydrocarbons in the apparatus described in A1.4.1.2. Charge the flask with 300 mL of sample, and place the temperature sensor such that the sensing point (bulb) is immersed in the sample. Heat the flask gently so that the oil temperature rises slowly to 150°C, vaporizing into the condenser by means of a small gas flame any drops of water that collect on the walls of the flask. Stop the distillation when no more water passes over in the distillate. Allow the residue to cool, run off the water collected in the separating funnel, and add the oily distillate back to the residue and mix thoroughly. If visible water is still present, filter by the method described in A1.5.1.2.

A1.6 Procedure

A1.6.1 Determine the density at 15°C of the crude petroleum by Test Method D 1298 or D 4052.

A1.6.2 Weigh the distillation flask to the nearest 0.1 g, and charge it with a mass equivalent to 100 mL of sample at 15°C

to within 0.1 g, ensuring that no oil flows down the vapor tube during this operation. Record the mass of oil (*D*). Place the liquid-in-glass thermometer so that the bottom of the bulb is 2.0 ± 0.5 mm from the bottom of the flask, or the alternative temperature sensor to a position established to give identical temperature readings.

NOTE A1.2—The position of the temperature sensor in the holder may be premarked before addition of the test portion to the flask.

A1.6.3 Swab out the condenser tube, and assemble the apparatus with the vapor tube of the flask extending into the condenser tube a distance of between 25 and 50 mm. Make the connection between flask and condenser tight by means of a stopper or bung through which the vapor tube passes.

A1.6.4 Place a clean dry receiver at the outlet of the condenser such that the condenser tube or adaptor extends into it at least 25 mm, but not below the 100-mL mark. Immerse the receiver up to the 100-mL mark in the transparent bath (see A1.4.1.1). Cover the top of the receiver closely during the distillation with a piece of paper weighted to restrict volatile losses and also to overcome the buoyancy of the receiver in the water bath. Circulate ice-cold (below 4°C) water through the condenser.

A1.6.5 Apply heat to the flask, and distil the crude petroleum at a rate not exceeding 2.5 mL/min until it ceases to foam, after which keep the rate of distillation constant at 2.0 to 2.5 mL/min (approximately 1 drop/s). Continue the distillation without interruption to 260°C. Remove the heat, and allow the condenser to drain into the receiver.

A1.6.6 Allow the residue in the flask to cool, remove the temperature sensor, and weigh the flask and residue. Record the mass of the residue (*R*).

A1.6.7 Use this residue to carry out the procedure (see Section 11) for the determination of asphaltene content.

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