



# Standard Test Method for Determination of Traces of Methanol in Propylene Concentrates by Gas Chromatography<sup>1</sup>

This standard is issued under the fixed designation D 4864; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon ( $\epsilon$ ) indicates an editorial change since the last revision or reapproval.

*This standard has been approved for use by agencies of the Department of Defense.*

<sup>ε1</sup> NOTE—Warning notes were placed in the text editorially in November 2000.

## 1. Scope

1.1 This test method covers the determination of methanol in propylene concentrates in the range of approximately 4 to 40 mg/kg (parts-per-million by weight).

1.2 The values stated in acceptable SI units are to be regarded as the standard.

NOTE 1—There is no direct acceptable SI equivalent for screw threads.

1.3 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.* Specific hazard statements are given in 11.1.1 and 12.11.

## 2. Referenced Documents

### 2.1 ASTM Standards:

D 4307 Practice for Preparation of Liquid Blends for Use as Analytical Standards<sup>2</sup>

E 260 Practice for Packed Column Gas Chromatography<sup>3</sup>

## 3. Terminology

### 3.1 Definitions:

3.1.1 *propylene concentrate*—concentrate containing more than 90 % propylene.

### 3.2 Definitions of Terms Specific to This Standard:

3.2.1 *outrag tube*—a length of 6.35-mm ( $\frac{1}{4}$  in.) outside diameter SS tubing normally attached to the inside end of a valve used on a pressure sampling cylinder. It is used to facilitate removal of a set quantity of liquified sample to prevent overpressuring the cylinder.

## 4. Summary of Test Method

4.1 A known weight of water is pressured into a sample

cylinder containing a known amount of liquified propylene. The contents in the cylinder are shaken and the water/methanol phase is withdrawn. A reproducible volume of the extract is then injected into a gas chromatograph (GC) equipped with either a thermal conductivity or a flame ionization detector. The methanol concentration is calculated from the area of the methanol peak using calibration and extraction factors obtained from synthetic blends of known methanol content.

## 5. Significance and Use

5.1 Methanol is a common impurity in propylene. It can have a deleterious effect on various processes that use propylene as a feedstock.

## 6. Interferences

6.1 There are no known interferences using the GC columns referenced in this test method. However, any water-soluble component that co-elutes with methanol on any other GC column used would interfere.

## 7. Apparatus

7.1 *Gas Chromatograph*—Any GC equipped with either flame ionization or thermal conductivity detectors with an overall sensitivity sufficient to detect at least 4 mg/kg of methanol.

7.2 *Column*—Any GC column that separates methanol from water, other alcohols, and any co-extracted hydrocarbons.

NOTE 2—See Table 1 for a suitable list of columns and Fig. 1 and Fig. 2 for examples of chromatograms. Also, refer to Practice E 260 for typical instructions in preparing such columns. Alternatively, columns can be purchased from commercial sources.

7.3 *Data Handling System*—Any commercially available GC integrator or GC computer system capable of accurately integrating the area of the methanol peak is satisfactory.

7.4 *Recorder*—A strip-chart recorder with a full scale response of 2 s or less and a maximum noise rate of plus or minus 0.3 % full scale.

7.5 *Sample Cylinders*, 300-mL capacity, stainless steel, Type DOT 3E (12409 kPa ((1800 psig)) working pressure).

7.6 *Balances*—Any types capable of weighing a 300-mL

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<sup>2</sup> *Annual Book of ASTM Standards*, Vol 05.02.

<sup>3</sup> *Annual Book of ASTM Standards*, Vol 14.02.

TABLE 1 Suitable Gas Chromatographic Columns and Temperatures<sup>A</sup>

Column Number	Column Size, m × mm	Tubing Type	Packing	Coating, μm Thickness	Oven Temperature, °C
1	1.22 × 6.35 OD	SS	15 % Carowax 1540 on 60/80 Chromosorb W AW	...	90
2	3.05 × 4.76 OD	SS	80/100 mesh Porapak QS	...	100
3	3.05 × 6.35 OD	Cu	10 % Carbowax 1540 on 30/60 mesh Chromosorb T	...	120
4	6.10 × 6.35 OD	Cu	10 % Carbowax 1540 on 30/60 mesh Chromosorb T	...	120
5	1.83 × 2 ID	glass	10 % Carbowax 20 M on 80/100 Chromosorb W AW	...	70
6	15 × 0.53 ID	fused silica	...	J&W DB-5, 1.5	70 to 120 at 2°/min

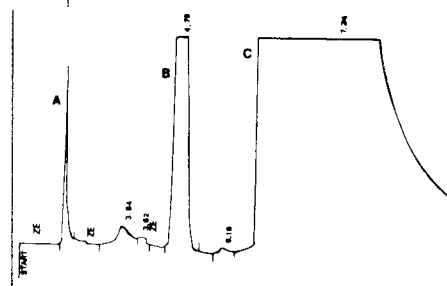
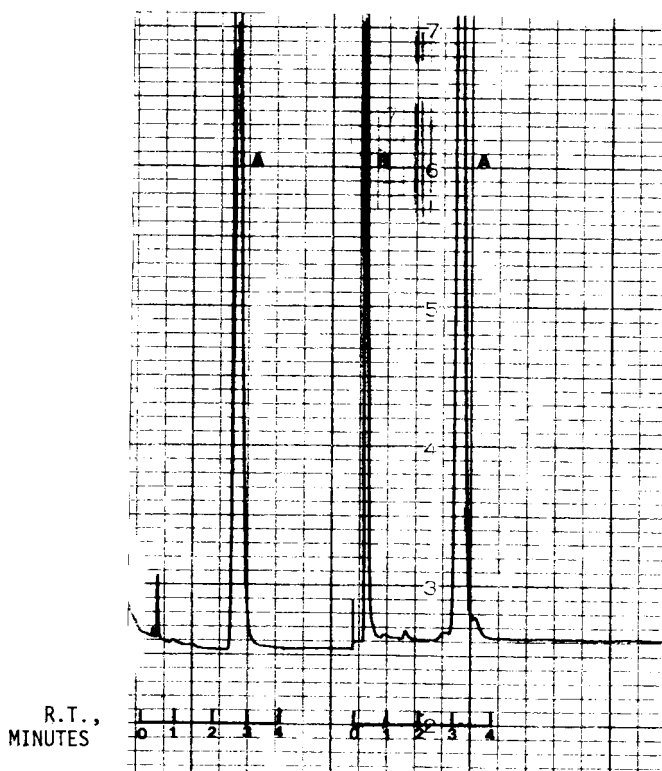
<sup>A</sup>These six columns have been tested cooperatively and have been found suitable for use with this test method.

**PEAK IDENTIFICATION**

- A METHANOL
- B PROPYLENE

**PEAK IDENTIFICATION**

- A UNKNOWN (PROBABLY PROPYLENE)
- B METHANOL
- C WATER



NOTE 1—Column used: No. 4 of Table 1; detector: thermal conductivity.

FIG. 2 Chromatogram of Water/Methanol/Propylene Extract

- 7.11 Hex Coupling, SS, ¼-in. female NPT by 30 mm (1.2 in.) long.
- 7.12 Brass Cap, ¼-in. NPT or optionally, a tube fitting nut, 6.35 mm outside diameter (¼in.). (See Note 3.)
- 7.13 Septum, TFE-fluorocarbon lined, 11-mm diameter.
- 7.14 Syringes, 10 and 25 μL.

**8. Reagents and Materials**

- 8.1 Methanol, reagent grade or better.
- 8.2 Propylene, 92+ % purity containing <0.2 mg/kg (ppmw) methanol.

**9. Sampling**

9.1 The propylene sample shall be in the liquified state and be representative of the material in the storage tank or process line. Also, for purposes of this method as well as for safety considerations, there must be a vapor space of about 15 % in the sampling container. It is recommended that sampling cylinders of the type listed in Section 7 be used. They can be equipped with an outage tube to effect the 15 % vapor space requirement.

**10. Preparation of Apparatus**

10.1 Prepare a water injection device. A suitable device is shown in Fig. 3. However, any other device that will deliver from 8 to 15 g of water can be used.

10.2 Prepare a 300-mL sample cylinder for use as a methanol cylinder, as shown in Fig. 4. (This cylinder must not

NOTE 1—Column used: No. 5 of Table 1; detector: flame ionization.  
 FIG. 1 Chromatograms of Water/Methanol Standard and Water/Methanol/Propylene Extract<sup>A</sup>

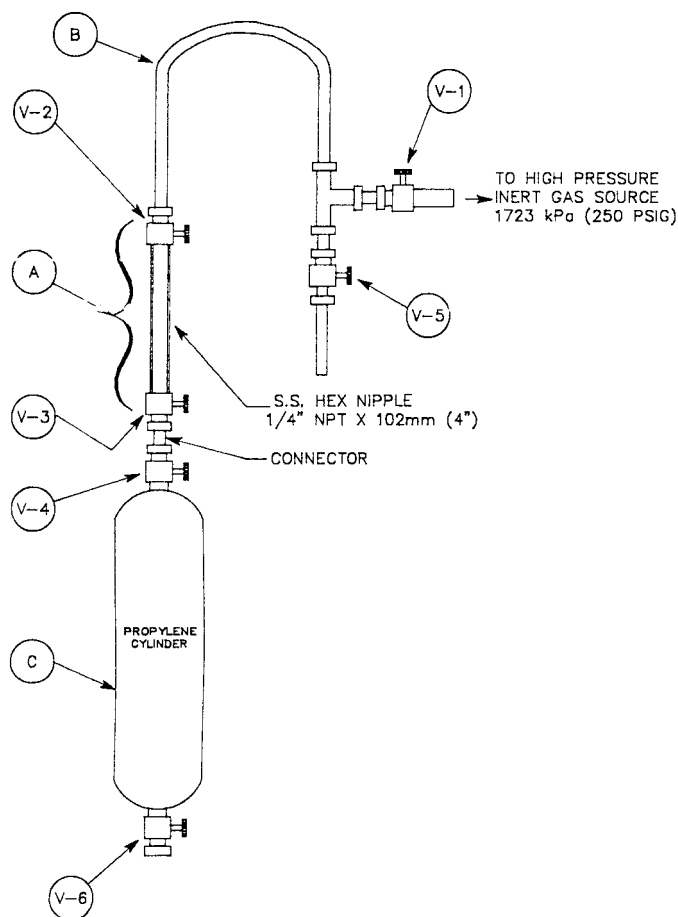
sample cylinder and contents accurately to 0.1 g and a 25-mL volumetric flask and contents accurately to 0.0001 g.

7.7 Plug Valve, ¼-in. male NPT or optionally, ¼-in. male NPT to 6.35 mm outside diameter (¼ in.) tubing. (See Note 3.)

7.8 Shut-off Valves, ¼-in. male NPT to 6.35 mm outside diameter (¼ in.) tubing.

7.9 Regulating Valves, ¼-in. male NPT and ¼-in. male NPT to ¼-in. female NPT.

7.10 Hex Nipple, SS, ¼-in. male NPT by 102 mm (4 in.) long.



- (A) Water injection device
- (B) 6.35 mm OD (1/4 in.) SS tubing
- (C) Sample cylinder containing propylene
- (V-1) & (V-5) Shut off valves, 1/4 in. male NPT to 6.35 mm OD (1/4 in.) tubing
- (V-2) & (V-3) Regulating valves, 1/4 in. male NPT to 1/4 in. female NPT
- (V-4) & (V-6) Regulating valves, 1/4 in. male NPT to 1/4 in. male NPT

FIG. 3 Water Injection Assembly

contain an outage tube.) Drill a 3 to 4 mm (approximately 1/8 in.) hole in a 1/4-in. NPT brass cap, insert an 11-mm septum into it, and screw it onto the plug valve.

NOTE 3—As an alternative, the cylinder can be equipped with 1/4-in. male NPT to a 6.35-mm (1/4-in.) outside diameter tubing plug valve. Then a 6.35-mm tube fitting nut can be used with the septum, thus avoiding the necessity of drilling a brass cap.

10.3 Set up the chromatograph in accordance with the manufacturer's recommendations. Install the analytical column and adjust the gas flows and temperatures so that methanol will elute at the desired time. Condition the column at operating conditions until a stable baseline is recorded at the required sensitivity.

## 11. Calibration

11.1 *Determination of Methanol Response Factor*—Prepare several aqueous solutions of methanol in the same concentration range as expected for samples to be analyzed.

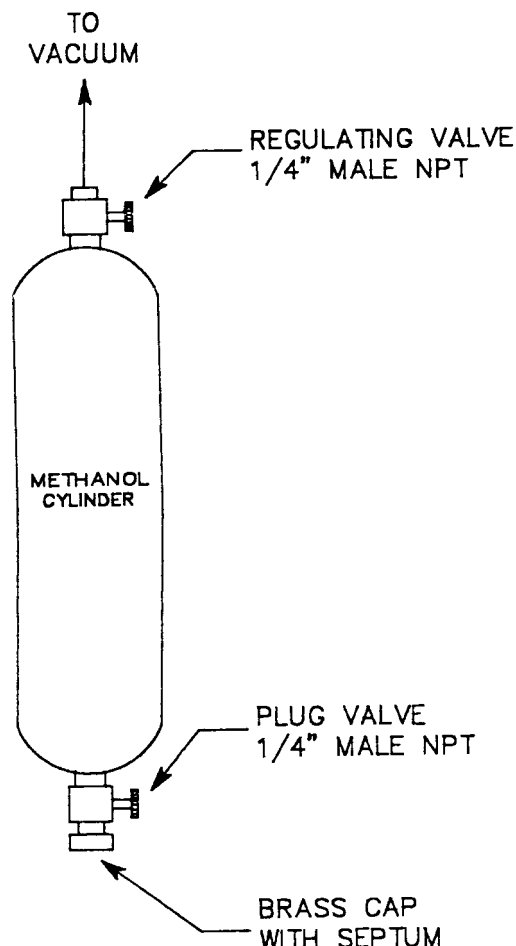


FIG. 4 Methanol Cylinder Extraction Factor Determination

NOTE 4—This should be approximately 40 to 400 mg/kg (ppmw) on the basis of propylene sample sizes of 100 to 120 g, water extract volumes of about 10 g, and methanol concentrations in the propylene of 4 to 40 mg/kg.

11.1.1 *Methanol Stock Solution*—Weigh an empty volumetric flask of at least 25 mL capacity to the nearest 0.0001 g. Add 20 mL of deionized water to the flask and reweigh. Finally, add 2 mL of methanol and again reweigh. Stopper and mix thoroughly. This should contain approximately 73 000 mg/kg (ppm by weight) of methanol. Calculate the exact concentration from the actual weights used. (**Warning**—Methanol is toxic and flammable. Use with adequate ventilation and keep away from ignition sources.)

NOTE 5—Refer to Practice D 4307 for additional information in preparing this solution and the calibration solution in 11.1.2.

11.1.2 *Methanol Calibration Solutions*—In similar manner, make serial dilutions by weight until two different concentrations in the range from 40 to 400 mg/kg are prepared.

11.1.3 With the GC at the proper operating conditions, inject an appropriate quantity of each calibration solution, in duplicate, and obtain the area of the methanol peak.

NOTE 6—The quantity of solution to be injected will depend largely on the type of detector in use. It varies from about 3 µL for FIDs to 25 µL for TCDs.

11.1.4 For each solution, calculate the response factor for methanol as follows:

$$F = C/H \quad (1)$$

where:

- $F$  = methanol response factor,
- $C$  = concentration of methanol, mg/kg, in the blend, and
- $H$  = area of the methanol peak (average of duplicate injections).

11.1.5 When the response factors at the two concentrations agree within 5 %, average them for use in the final calculation given in Section 13.

11.2 *Determination of Methanol Extraction Factor*—Since the methanol is not extracted quantitatively due to solubility competition between the water and the propylene, the extraction efficiency must be determined experimentally as follows.

11.2.1 Collect 100 to 120 g (190 to 230 mL) of methanol-free propylene in a tared 300-mL sample cylinder. Reweigh the cylinder to ensure that it contains the proper amount.

NOTE 7—Liquified propylene is at high pressure, can cause frostbite, and is flammable. Use appropriate care in handling.

11.2.2 As shown in Fig. 4, attach the opposite end of the septum equipped methanol cylinder to a vacuum source. Be sure that this assembly is leak-free.

11.2.3 Open both valves and evacuate the cylinder up to the septum. Then close the plug valve (next to the septum) and continue the evacuation. Finally, close the other cylinder valve, disconnect the cylinder from the vacuum source, and weigh it to the nearest 0.1 g.

11.2.4 Flush a 10 or 25- $\mu$ L syringe several times with methanol, then fill it to the desired volume (see Table 2), wipe off the tip, pull the plunger back about 1  $\mu$ L, and weigh it to the nearest 0.0001 g.

11.2.5 Open the cylinder plug valve at the septum and immediately inject the methanol in the syringe through the septum into the cylinder. Close the valve and immediately reweigh the syringe to determine the amount of methanol injected. The difference between this weight and that obtained in 11.2.4 is the weight of methanol injected,  $W_1$ .

11.2.6 Cool the evacuated cylinder to about 20°C below the temperature of the propylene cylinder.

11.2.7 As shown in Fig. 5, connect the cylinder containing propylene to the evacuated cylinder containing methanol via a hex coupling, a short length of 6.35-mm (1/4-in.) outside diameter SS tubing, or any other suitable fitting. Before tightening, flush the connector with a small amount of propylene by briefly opening the lower valve on the propylene cylinder.

11.2.8 With both cylinders in a vertical position and the propylene cylinder on top, open the valves between them (*propylene cylinder first*) and allow the liquified propylene to flow into the evacuated cylinder.

11.2.9 Close the valves, disconnect the cylinders, and allow

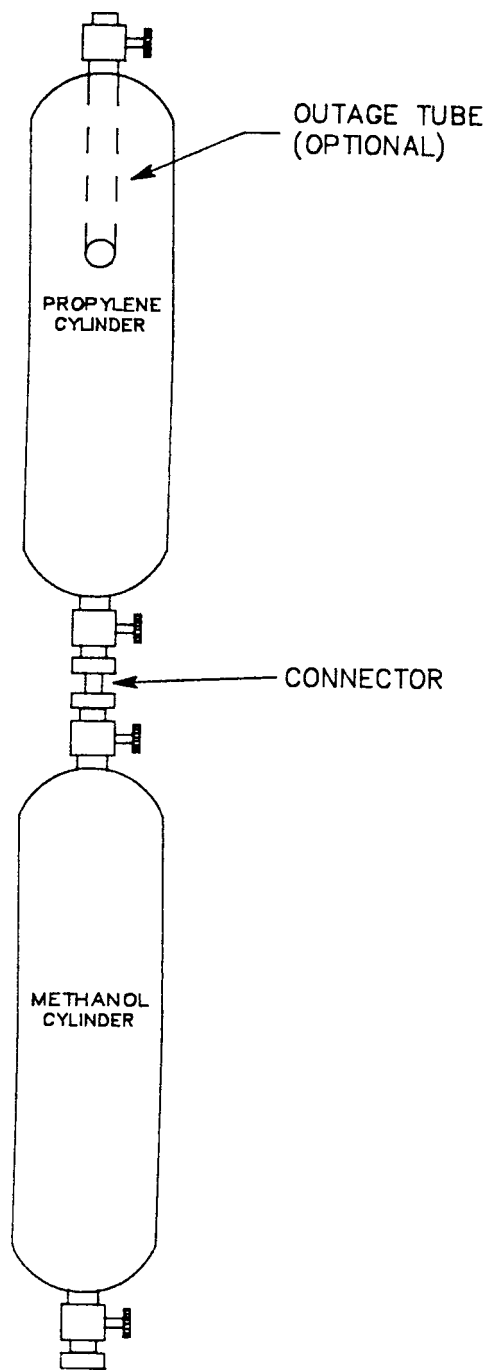


FIG. 5 Cylinders Assembly Extraction Factor Determination

the lower cylinder to warm to room temperature. Wipe off any water condensation and allow to dry.

11.2.10 Weigh the cylinder containing the methanol and propylene blend to the nearest 0.1 g. The difference between this weight and that obtained in 11.2.3 is the weight of methanol and propylene,  $W_2$ . Calculate the concentration of methanol as follows:

$$C = (W_1 \times 10^6)/W_2 \quad (2)$$

where:

$C$  = concentration of methanol, mg/kg,

TABLE 2 Methanol Injection Sizes

Methanol Volume, $\mu$ L	Weight, mg	Equivalent Concentration, mg/kg for 100 g propylene
5	4	36
10	8	72
15	12	108
20	16	144

$W_1$  = weight of methanol injected, g, and  
 $W_2$  = weight of propylene plus methanol, g.

11.2.11 Shake the cylinder vigorously to mix the propylene and methanol. Then extract the methanol and analyze the extract as described in 12.2-12.13. Analyze the extract in duplicate and average the methanol peak areas.

11.2.12 Calculate the methanol content of the extract as described in 13.1, but exclude  $F_x$ , the methanol extraction factor.

11.2.13 Calculate the methanol extraction factor as follows:

$$F_x = C/D \quad (3)$$

where:

$F_x$  = methanol extraction factor,  
 $C$  = methanol concentration calculated in Eq 2, and  
 $D$  = methanol concentration calculated in 11.2.11.

NOTE 8—It is recommended that the extraction procedure be repeated at a different concentration to verify the accuracy of the factor. Extraction factors of 1 to 2 are typical.

## 12. Procedure

12.1 Weigh the sampling cylinder containing at least 100 grams of propylene to the nearest 0.1 g.

NOTE 9—When practical, it is advisable to weigh the sampling cylinder before sampling to obtain a tare weight.

12.2 Pressure deionized water into the injection device in a vertical position from the bottom and close the valves.

12.3 As shown in Fig. 3, connect the injection device to the sample cylinder using a hex coupling or other suitable device.

12.4 Attach the other end to an inert gas source at 1724 kPa (250 psig) and purge the lines between  $V_2$  and  $V_5$ .

12.5 Close  $V_5$  and tighten the connection at  $V_2$ .

12.6 Pressure the water into the cylinder by opening Valves  $V_1$ ,  $V_2$ ,  $V_3$ , and  $V_4$ , in that order.

12.7 Close valves  $V_1$  and  $V_4$  and depressure the device via  $V_5$ . Then disconnect the cylinder.

12.8 Remove any residual water from the outlet of  $V_4$  and reweigh the cylinder to 0.1 g. The difference between this weight and that in 12.1 is the weight of the water,  $W$ .

12.9 Shake the cylinder vigorously for at least 10 min.

12.10 Clamp the cylinder in a vertical position and allow the aqueous phase to settle. (When the cylinder contains an outage tube, it must be at the top of the cylinder.)

12.11 Carefully open the bottom valve and drain the aqueous phase containing the methanol into an appropriate container (vial or flask) and cap it. (**Warning**—As soon as the aqueous phase drains, high-pressure liquified propylene will surge out.)

12.12 With the GC at the proper operating conditions, inject

the appropriate size of the aqueous extract for the GC and calibration being used.

12.13 Safely vent off the remaining propylene in the sample cylinder and reweigh it to 0.1 g (unless the cylinder was already tared.) The difference between this weight and that in 12.1 is the weight of the propylene extracted,  $G$ .

## 13. Calculation

13.1 Calculate the methanol content of the propylene using the following equation:

$$\text{Methanol, mg/kg (ppm wt)} = (A \cdot F \cdot F_x \cdot W) / G \quad (4)$$

where:

$A$  = area of methanol peak,  
 $F$  = methanol response factor for the sample size used (Eq 1),  
 $F_x$  = methanol extraction factor (Eq 3),  
 $W$  = weight of water used in the extraction, g, and  
 $G$  = weight of propylene extracted, g.

## 14. Precision and Bias <sup>4</sup>

14.1 *Precision*—The precision of this test method as determined by the statistical examination of interlaboratory test results is as follows:

14.1.1 *Repeatability*—The difference between successive results by the same operator with the same apparatus under constant operating conditions on identical test material would, in the long run, in the normal and correct operation of the test method exceed the following value only in one case in twenty:

Repeatability = The maximum allowable ratio of the larger to the smaller result is 2.4.

14.1.2 *Reproducibility*—The difference between two single and independent results obtained by different operators working in different laboratories on identical test material would, in the long run, exceed the following values only in one case in twenty:

Reproducibility = The maximum allowable ratio of the larger to the smaller is 8.0.

14.2 *Bias*—Since there is no accepted reference material suitable for determining the bias for this procedure for measuring methanol, the bias is not available for this test method.

## 15. Keywords

15.1 gas chromatography; methanol; propylene

<sup>4</sup> The values in the statements were determined in a cooperative program following RR: D02-1007. The data of the program are filed at ASTM Headquarters. Request RR: D02-1243.

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